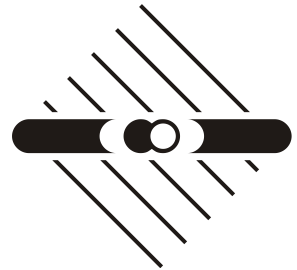


Liquid scintillator R&D in Europe

Christian Buck, MPIK Heidelberg

NNN08

Paris, September, 2008



MAX-PLANCK-GESELLSCHAFT

Overview

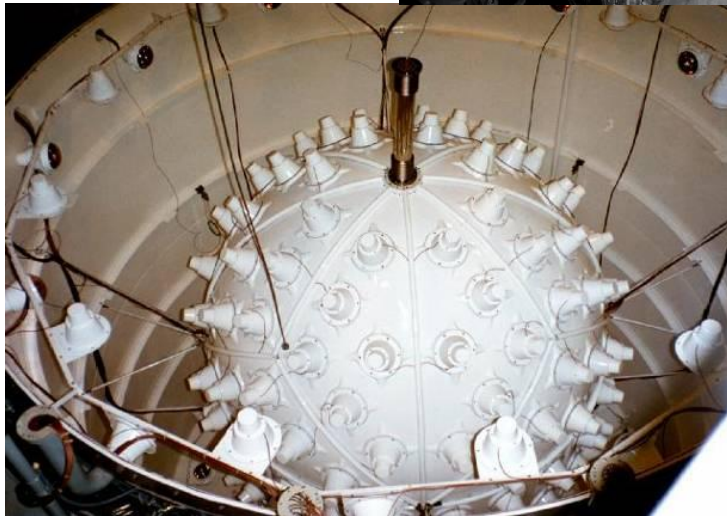
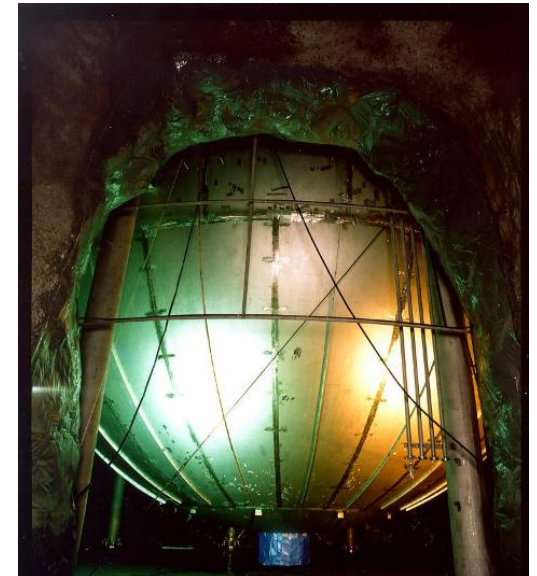
- Introduction
- Scintillator components
- Energy transfers
- Metal loaded scintillators
- Conclusions



Liquid scintillator past and present

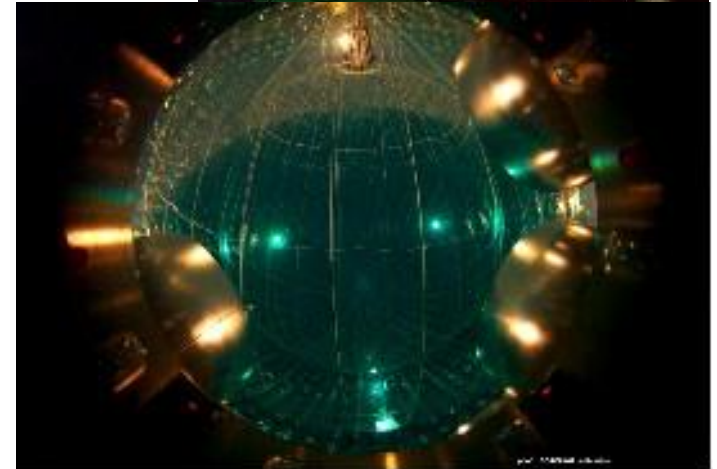
Metal loaded:

- Reines
- Bugey
- Chooz
- Palo Verde



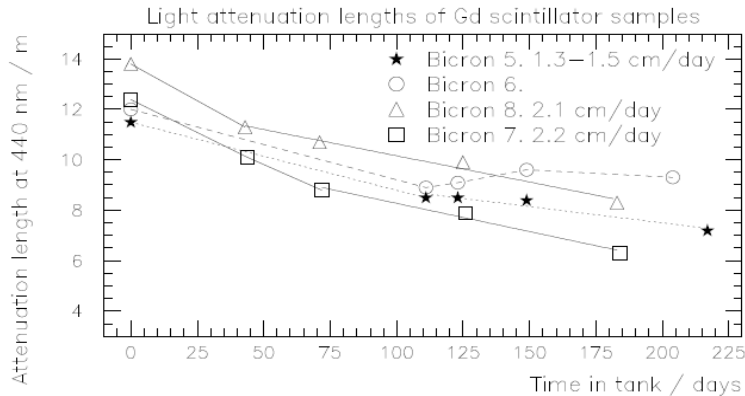
Unloaded:

- KamLand
- Borexino



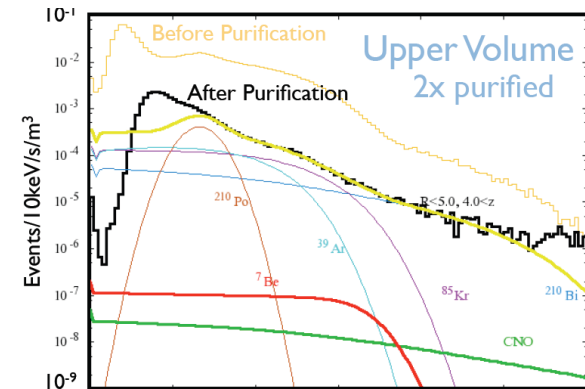
Challenges: Stability and purity

Palo Verde:



A.G.Piepke, S.W.Moser, V.M.Novikov; NIM A 342 (1999) 392-398

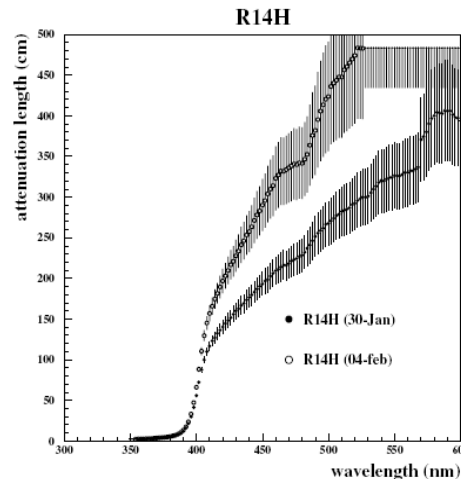
KamLAND Background:



KamLAND Collaboration, Neutrino 2008, Christchurch, NZ

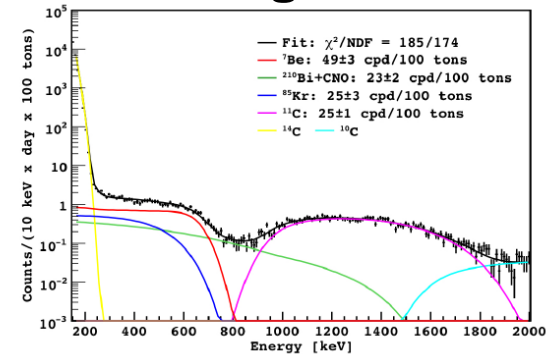
Chooz:

$Gd(NO_3)_3$
 $\tau \sim 240$ days



Chooz Coll.; Eur.Phys.C27, 331-374 (2003)

Borexino Background:



Borexino Collaboration, arXiv:0805.3843v2 [astro-ph], Jun 2008

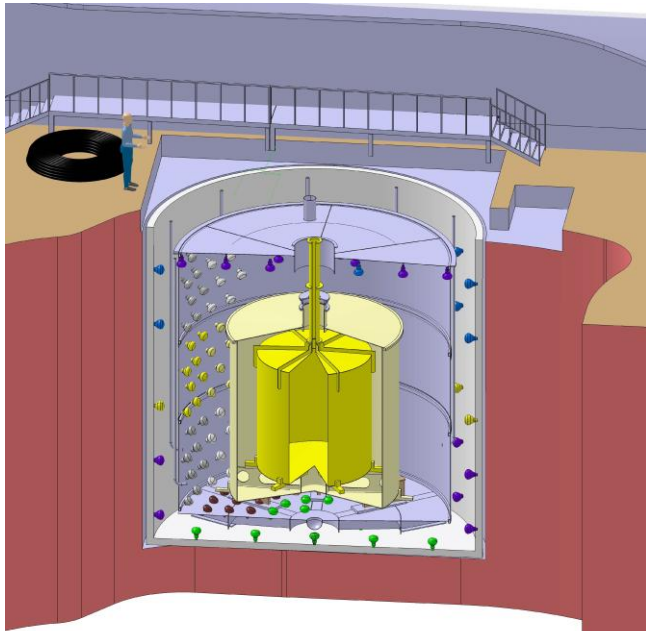


Liquid scintillator properties

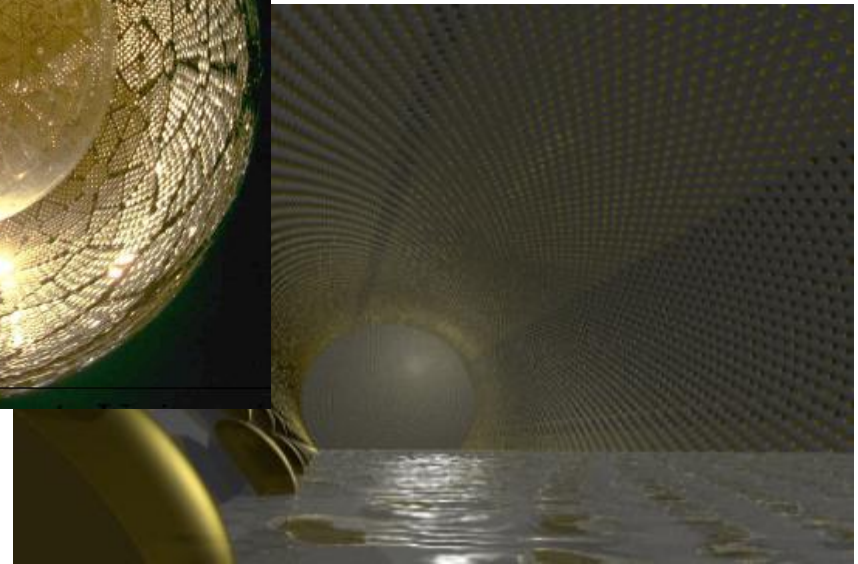
- High energy resolution
- Low energy detection threshold
- high purity (Borexino)
- fast signals (better understanding of timing properties)
- moderate cost
- Improved stability of metal loaded scintillators



Liquid scintillator future

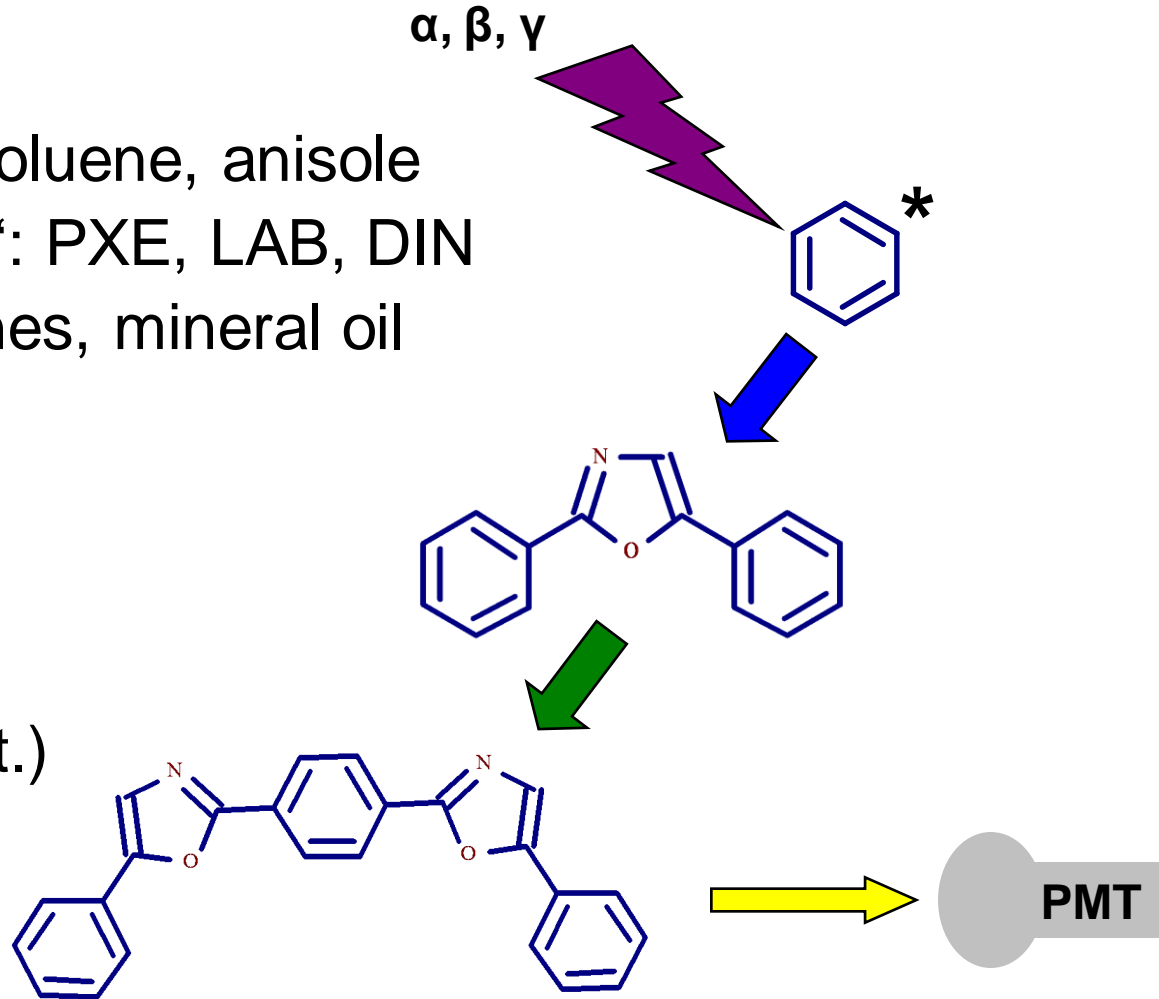


- Double Chooz
- Daya Bay
- SNO+
- LENA
- ...



Liquid scintillator components

- Solvent
 - pseudocumene, toluene, anisole
 - „Safe scintillators“: PXE, LAB, DIN
 - Admixtures: alkanes, mineral oil
- Primary fluor
 - PPO
 - BPO
 - Butyl-PBD,...
- Secondary fluor (opt.)
 - Bis-MSB
 - POPOP



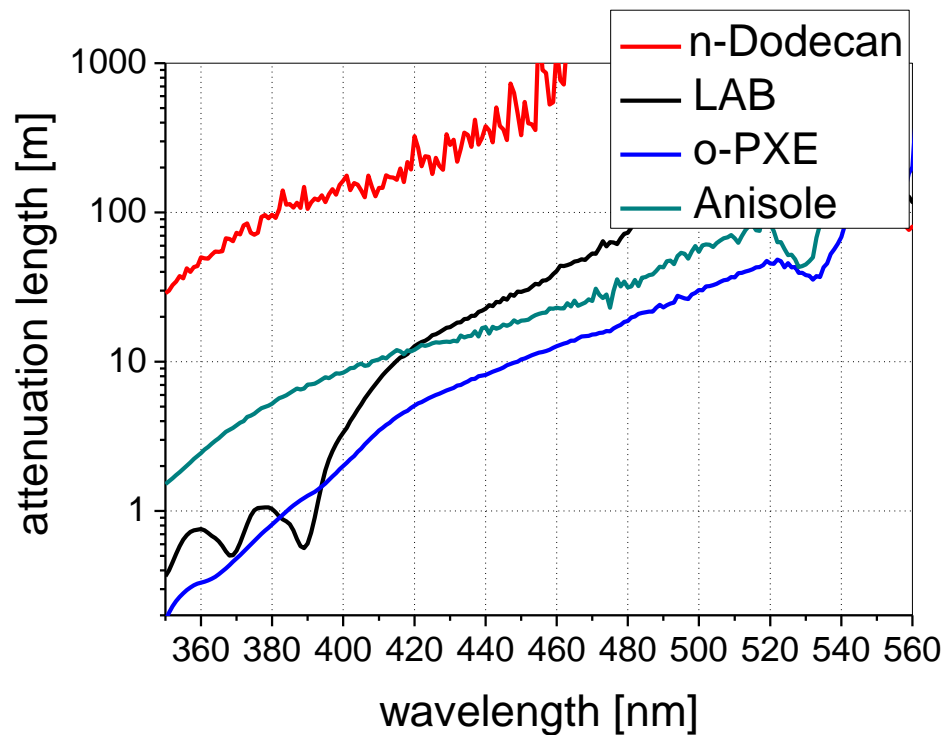
Comparison solvents

Scintillation yield

solvent	Yield
PC	1.00
Anisole	0.81
PXE	0.88
LAB	0.74
n-Dodecan	0.40
Oil	0.33

MPIK measurements (6 g/l PPO)

Attenuation length

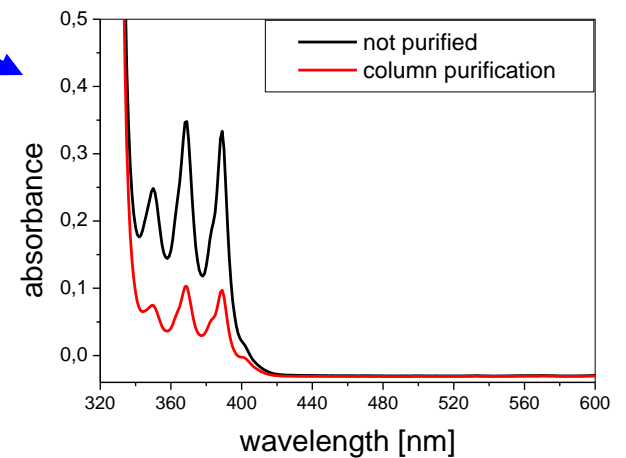
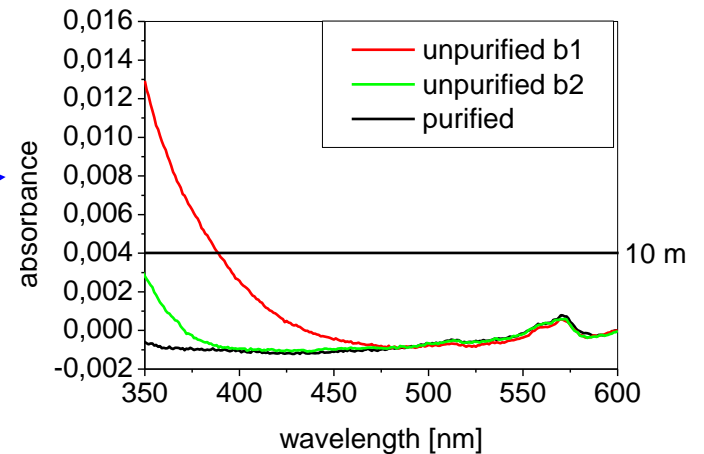


- UV/Vis (MPIK): (absorbance + scattering)
- Single contributions determined at TU Munich



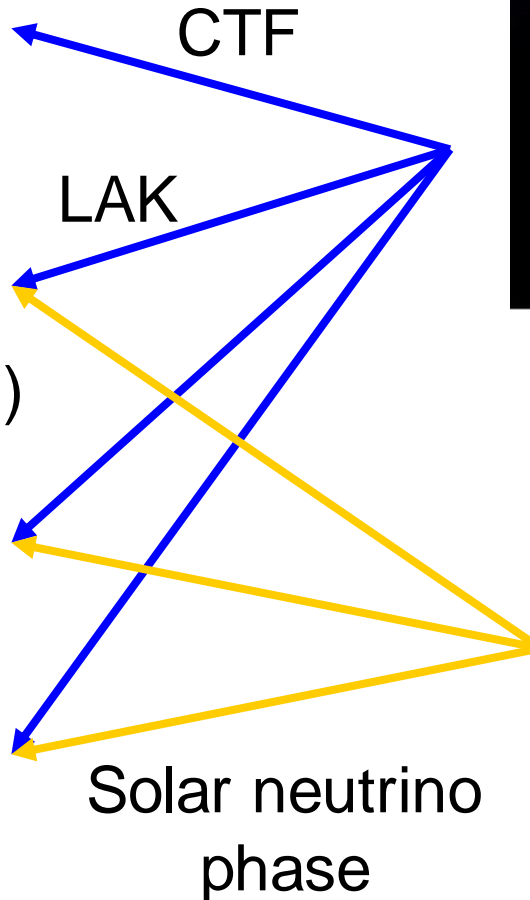
Purification methods

- Column purification
 - Radioimpurities
 - Optics
- N₂ purging
 - Radon, ⁸⁵Kr
 - Light yield (oxygen)
- Water extraction
 - Radioimpurities (e.g. ⁴⁰K)
- Distillation
 - Radioimpurities
 - Optics

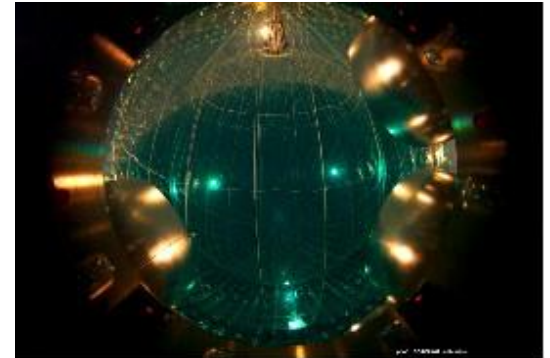


Purification methods

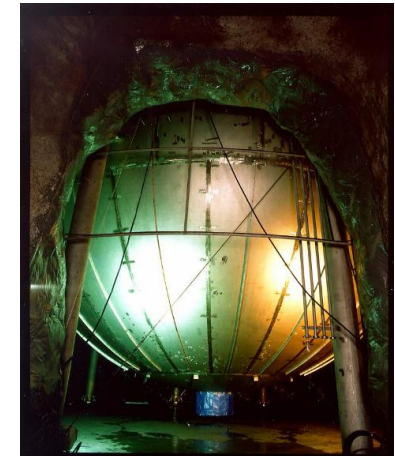
- Column purification
 - Radioimpurities
 - Optics
- N₂ purging
 - Radon, ⁸⁵Kr
 - Light yield (oxygen)
- Water extraction
 - Radioimpurities
- Distillation
 - Radioimpurities
 - Optics



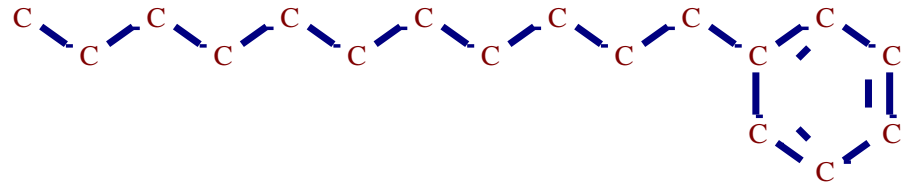
Borexino



KamLand



LAB



➤ Is LAB a new high light yield, transparent solvent?

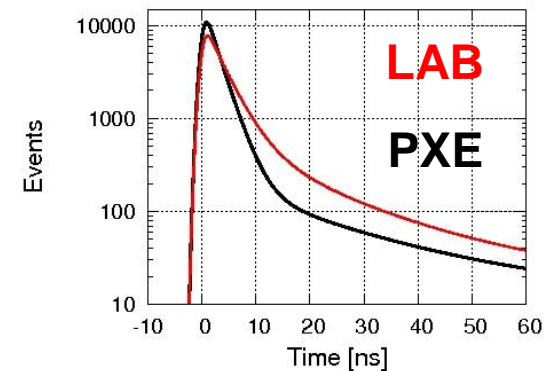
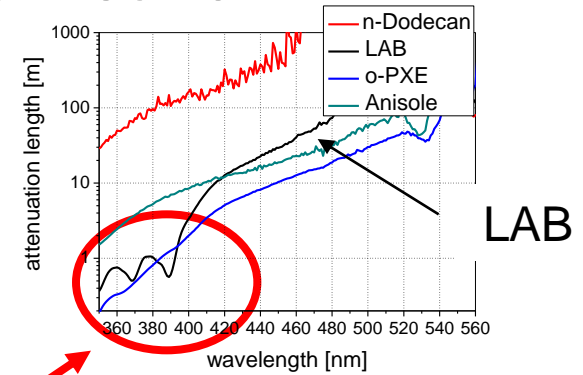
Not really.

- Used since decades
- Average light yield
- Average transparency

➤ It is a high flash point, low toxicity solvent at moderate cost and reasonable optics,

➤ ...but:

- Mixture
- Biphenyls (absorption/emission!)
- Timing properties



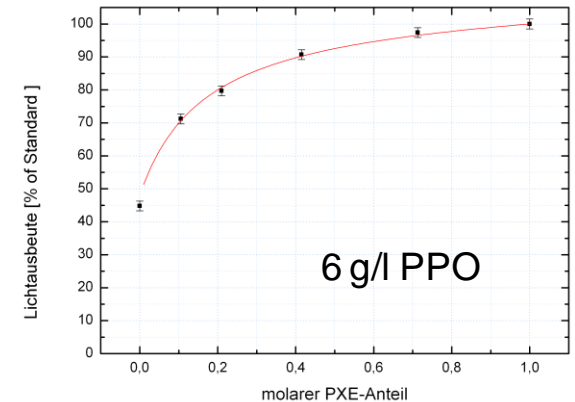
Solvent mixtures

Advantage: Parameters tuneable

- optimize material compatibility
- change timing properties
- match density
- adjust light yield

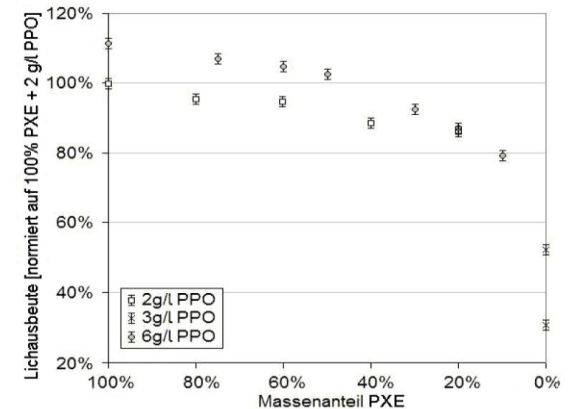
Light production in alkanes

- Radiation creates e^- - hole pairs
- Recombination, fragmentation, radicals, reactions → excited molecules
- → energy transfer to fluors



C.Buck, dissertation, MPIK (2004)

C.Aberle, diploma thesis, MPIK (2008)



M.Wurm, diploma thesis, TUM(2005)



Comparison fluors

PPO



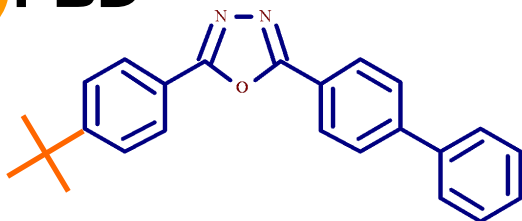
- transparent
- well established
- high quantum yield



(Butyl-)PBD



- high light yield

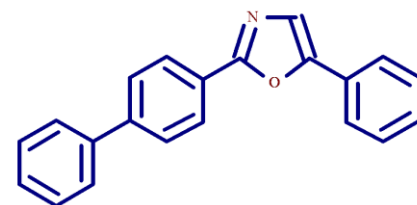


- poor quantum yield
- costs

BPO



- high(est) light yield
- emission around 400 nm

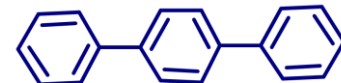


- absorption properties
- limited availability

pTP



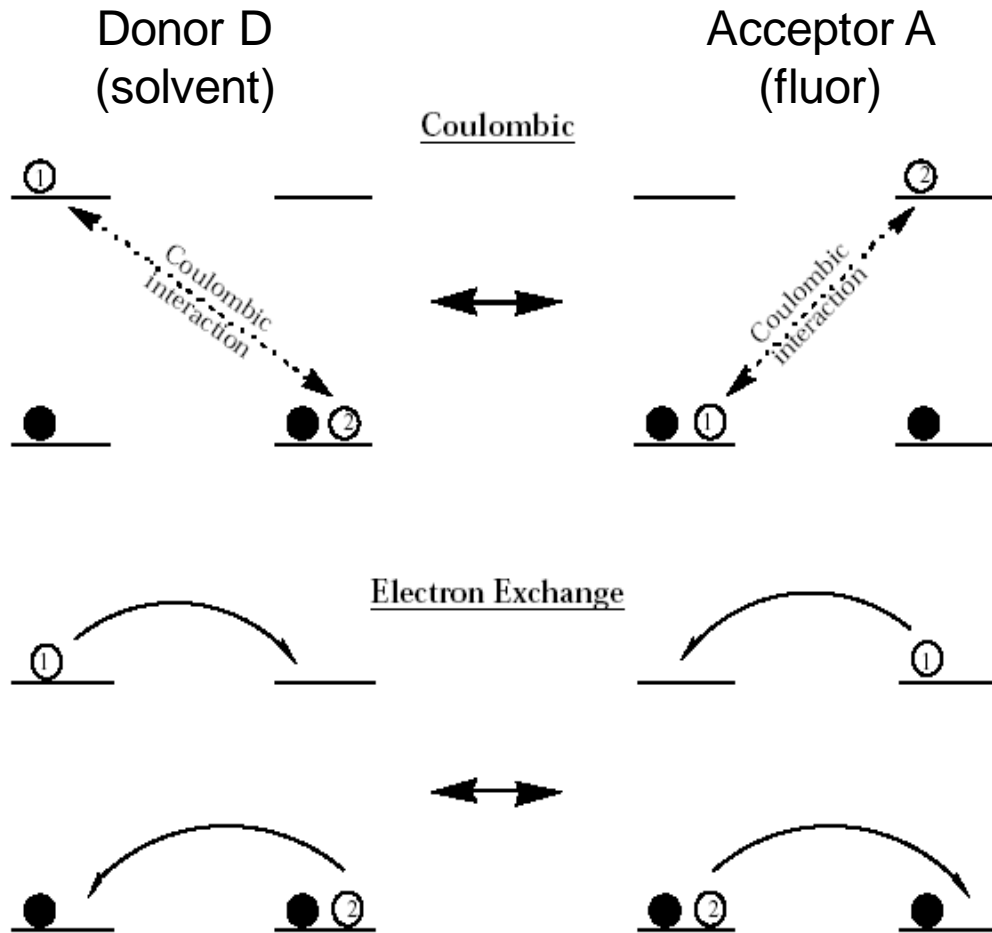
- fast
- overlap with bis-MSB



- low solubility
- poor quantum yield



Energy transfer (non-radiative)



$$k_{FRET} \propto \left(\frac{r_0}{R} \right)^6 \cdot J$$

$$J = \int \frac{\epsilon_A(\nu) \cdot f_D(\nu)}{\nu^4} d\nu$$

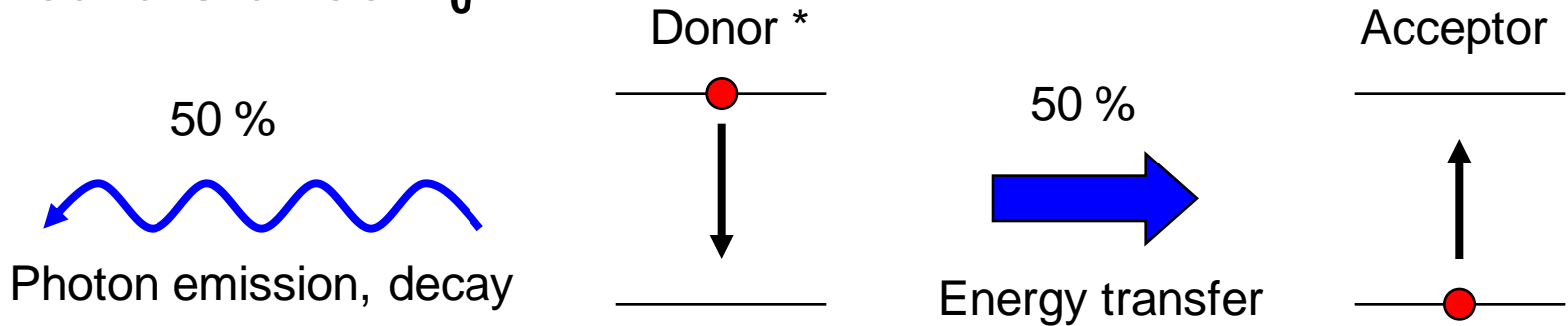
$$k_{exchange} \propto J \cdot e^{-2R/L}$$

T. Förster: Annalen der Physik 2 (1948) 55.
D.L. Dexter: J. of Chem. Lett. 21 (1952) 836.



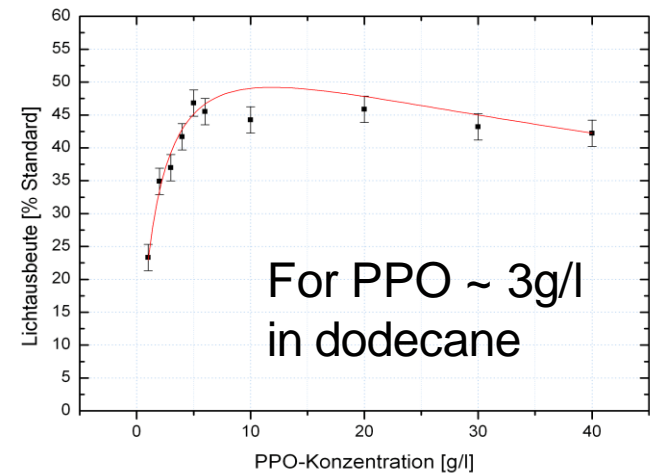
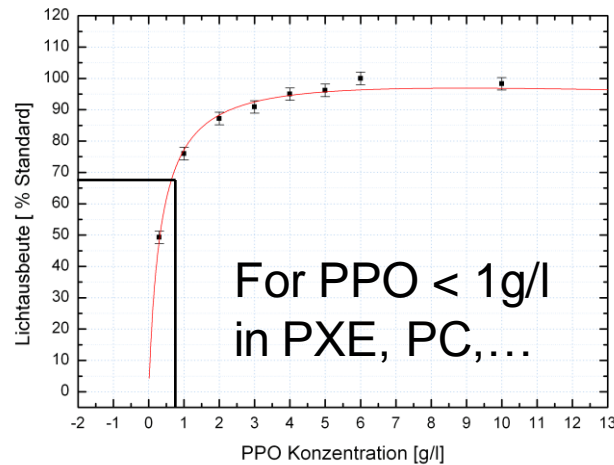
Critical concentration

Critical distance R_0 :

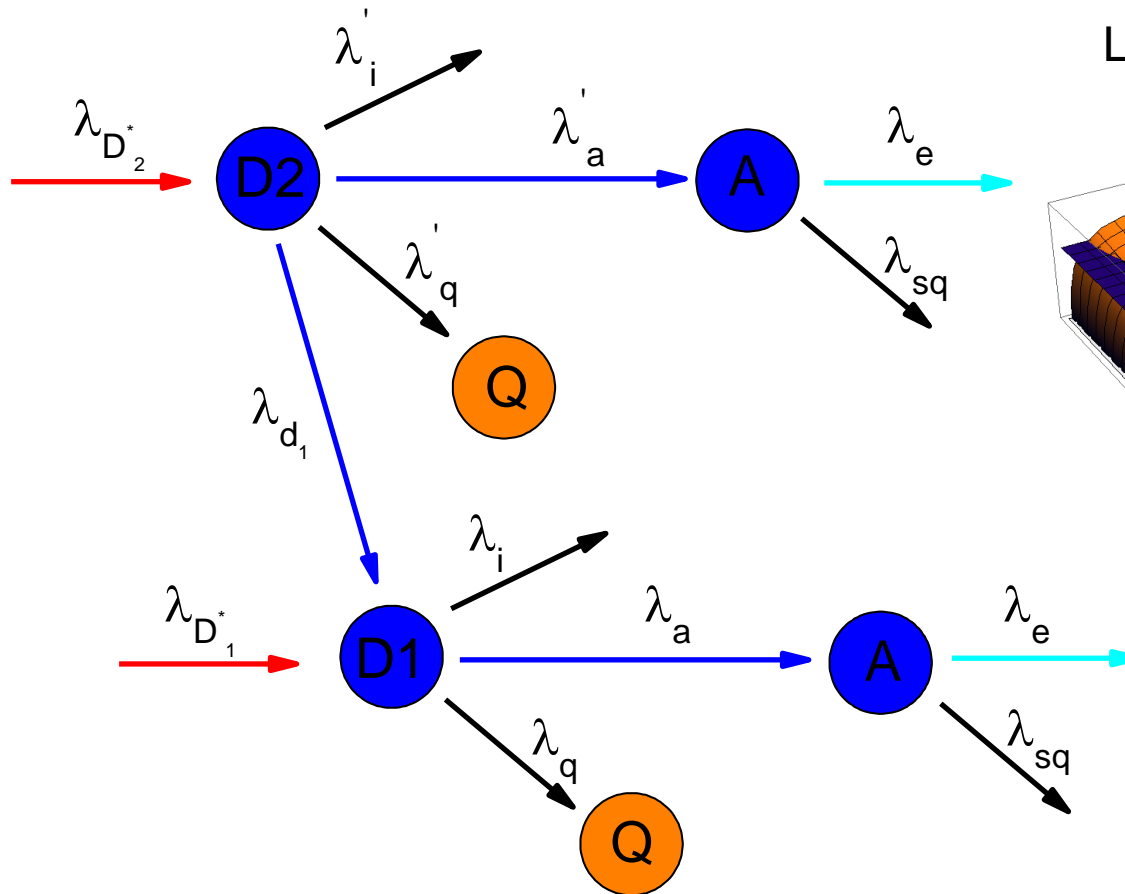


Critical conc.:

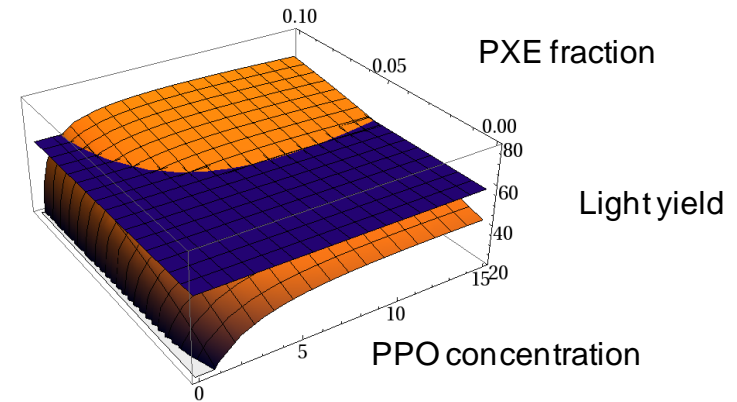
$$C_0 \propto \frac{1}{R_0^3}$$



Light yield model



Light yield predictions

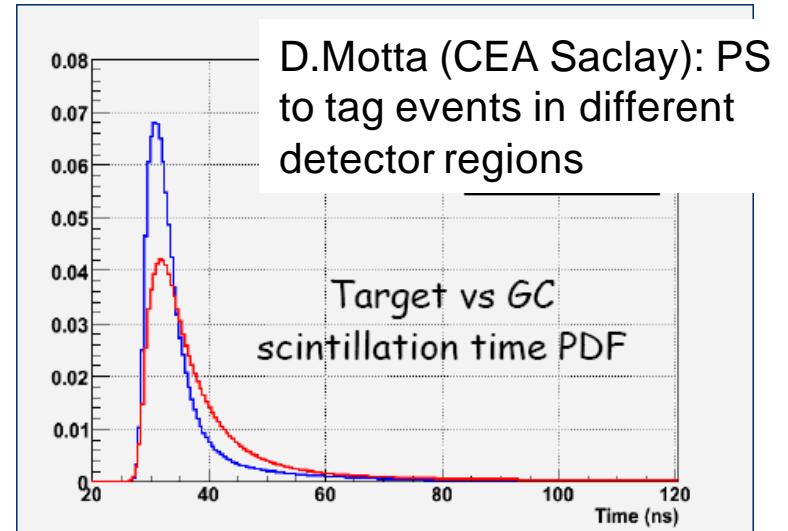
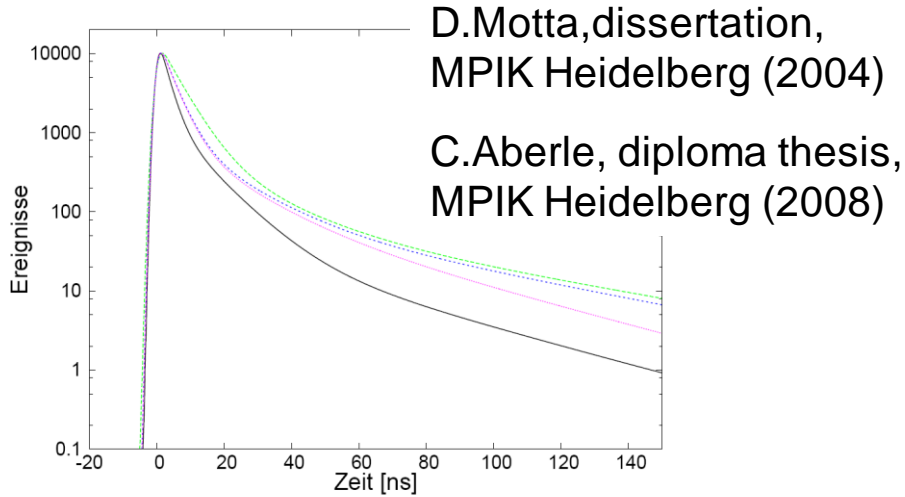


C.Buck, F.X. Hartmann,
D.Motta, S.Schönert, CPL,
435 (2007) 252 - 256

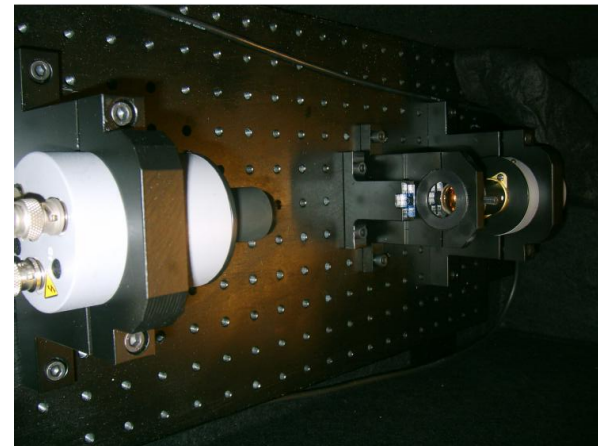
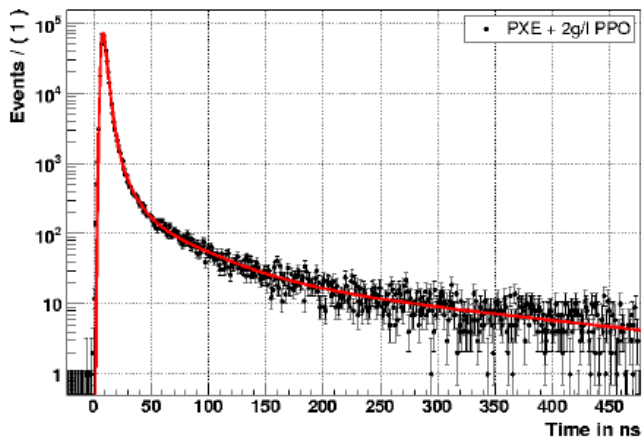
C.Aberle, diploma thesis,
MPIK Heidelberg (2008)



Timing properties

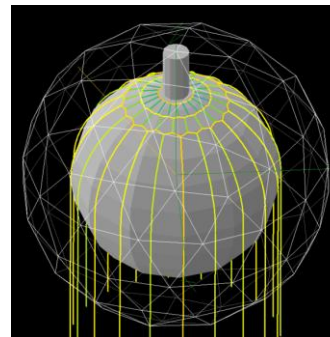
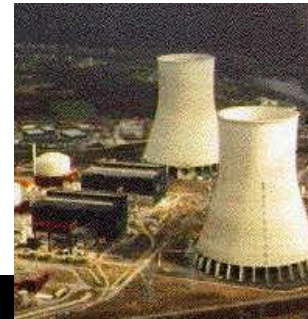
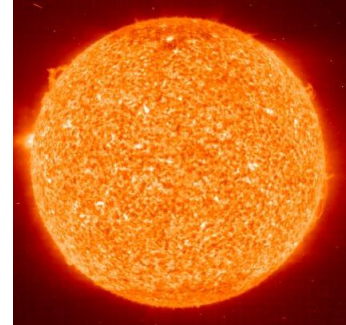


T. Marrodan Undagoitia, dissertation, TU München (2008)



Metal loaded scintillators

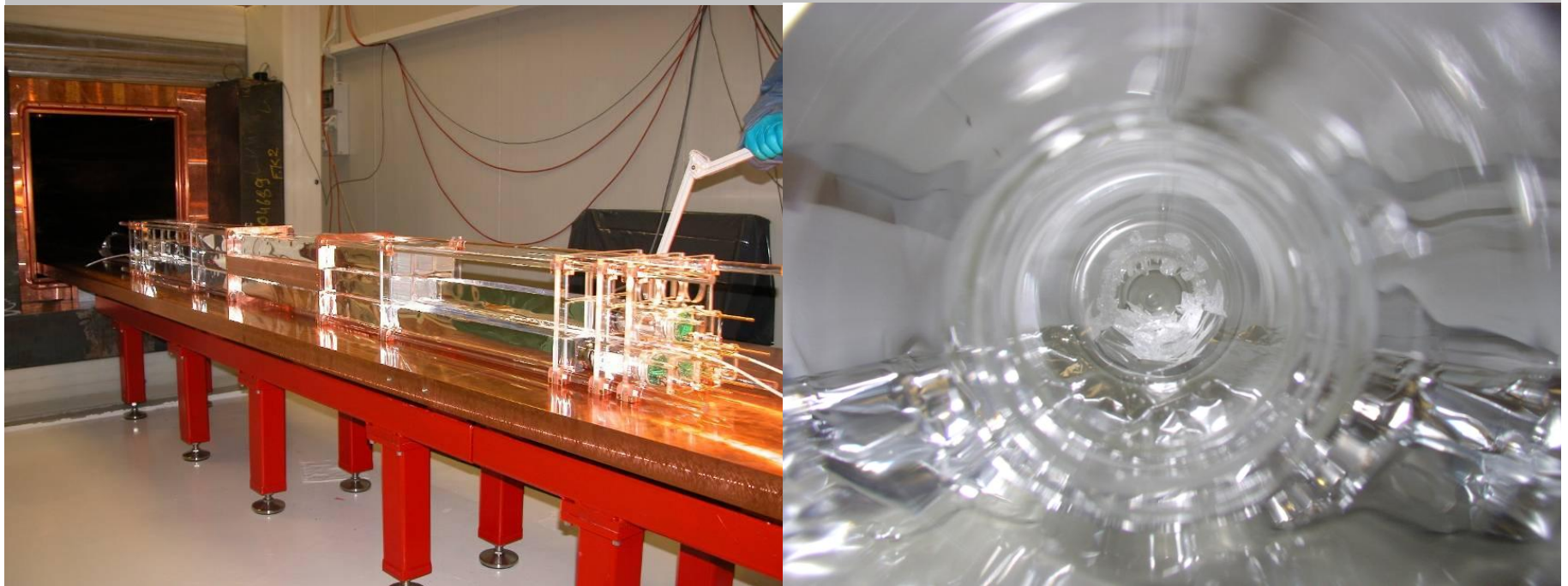
- Solar neutrinos (LENS, SIREN):
 - Metal: Ytterbium, Indium, Gadolinium
 - Challenge: High loadings
- Reactor (Double Chooz, Daya Bay)
 - Metal: Gadolinium
 - Challenge: Stability
- $\beta\beta$ -decay (SNO+)
 - Metal: Neodymium
 - Challenges: transparency; purity



Indium

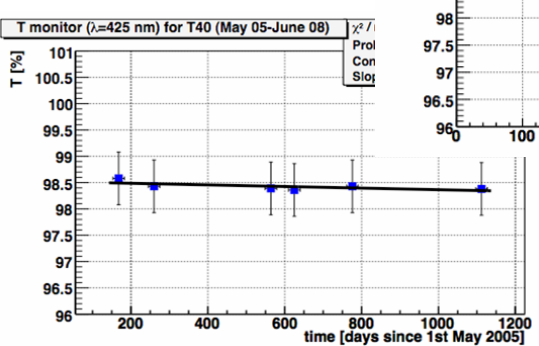
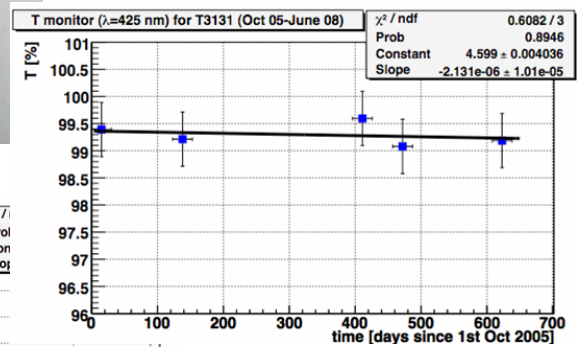
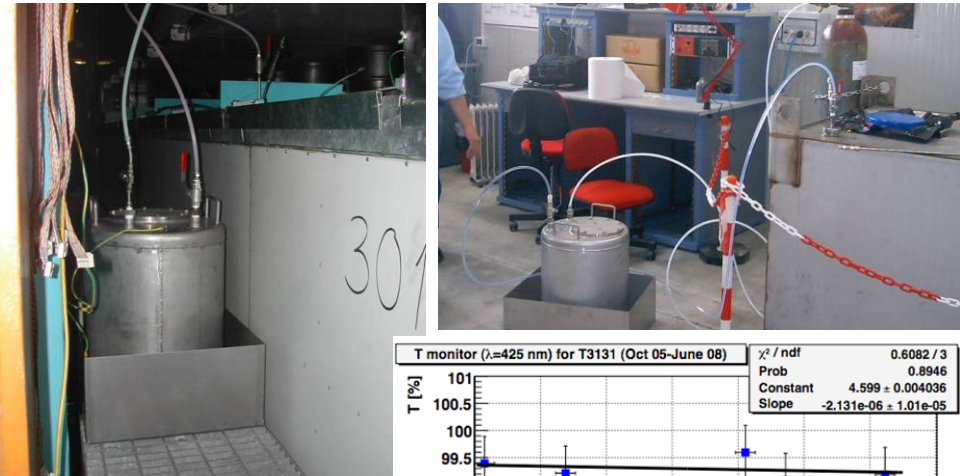
Indium-loaded scintillators at LLBF > 1 year

- MPIK: $\text{In}(\text{acac})_3$ (F.X.Hartmann et al.)
 - INR/LNGS: Carboxylic acid version
- } > 50 g/l Indium



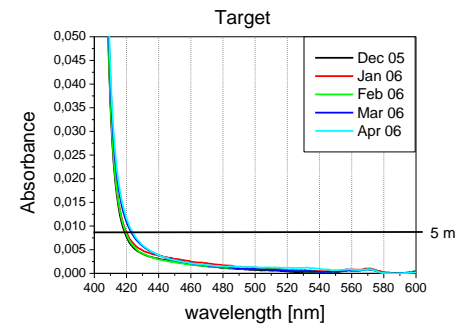
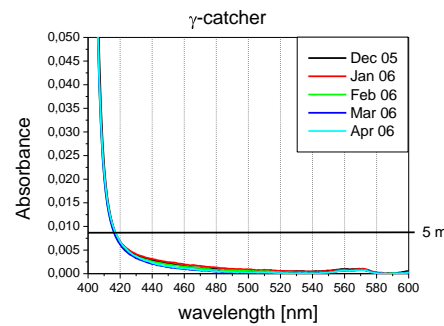
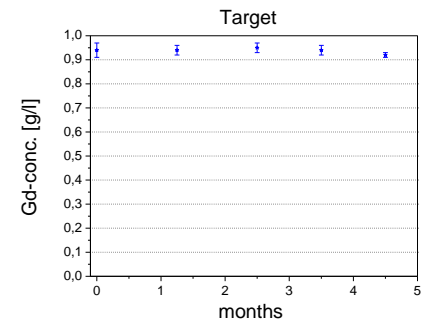
Gadolinium (Carboxylates)

**INR/LNGS: 2 x 1.2 t Gd-LS
(0.1%) in frame of LVD**

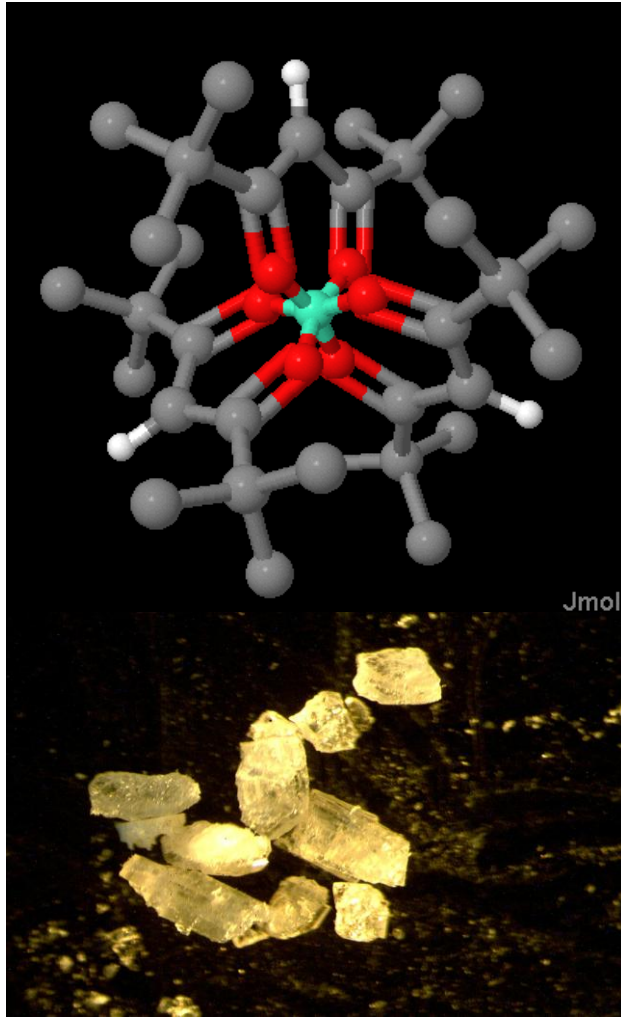


**Stability over 2 y
in SS tanks**

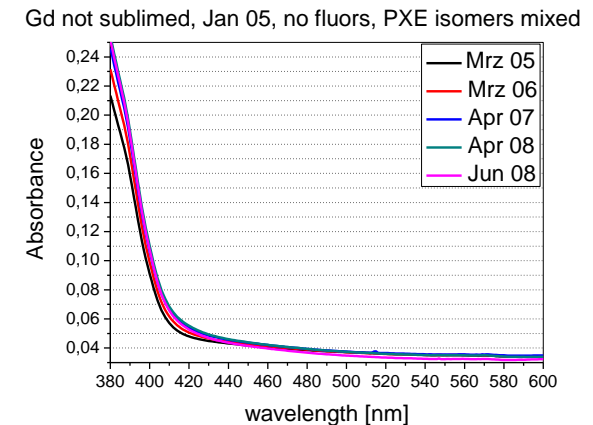
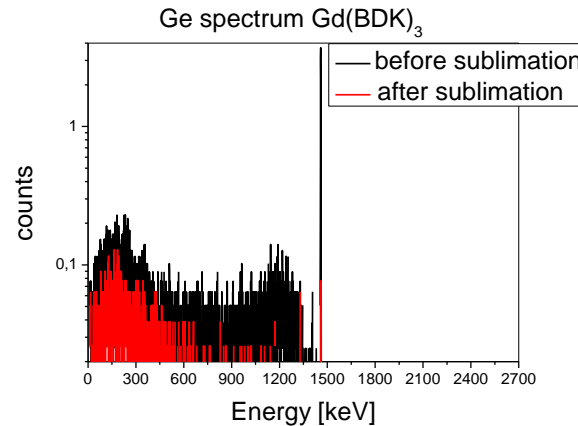
**Double Chooz mockup
(TMHA, MPIK Hd 2003):**



Gadolinium (β -diketones)



- Purified by sublimation
- stability tests at MPIK since > 3 years
- att. length (1 g/l) > 50 m above 420 nm
- 100 kg produced (2 DC detectors)
- stability/compatibility at CEA Saclay

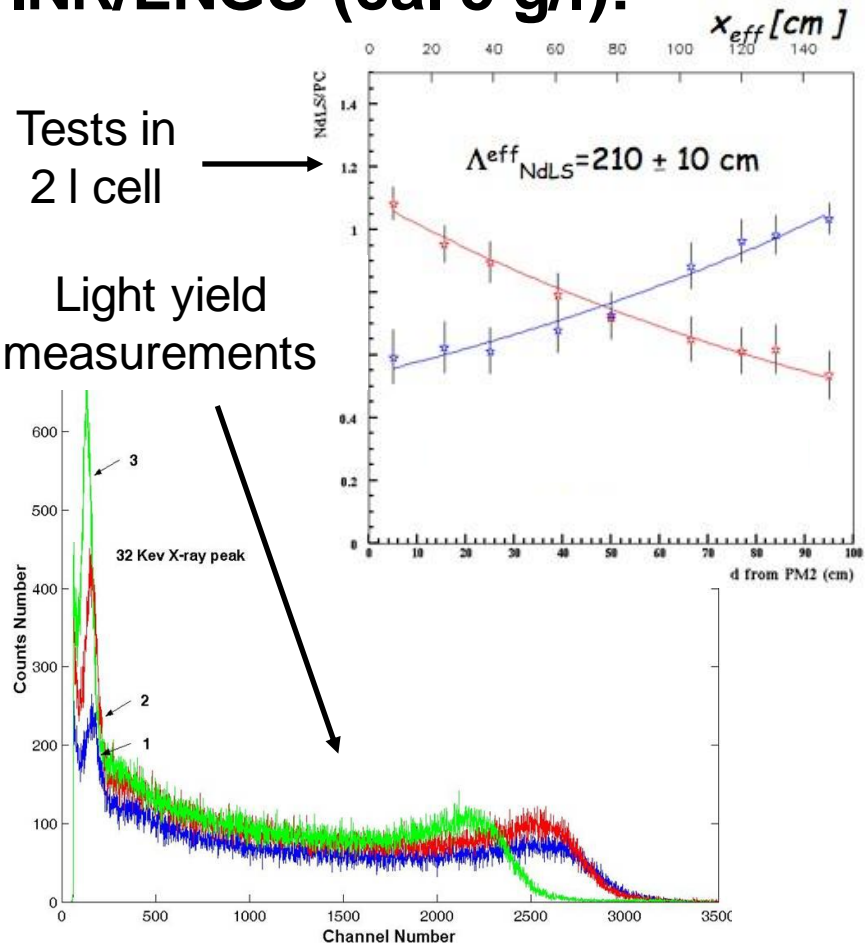


Neodymium

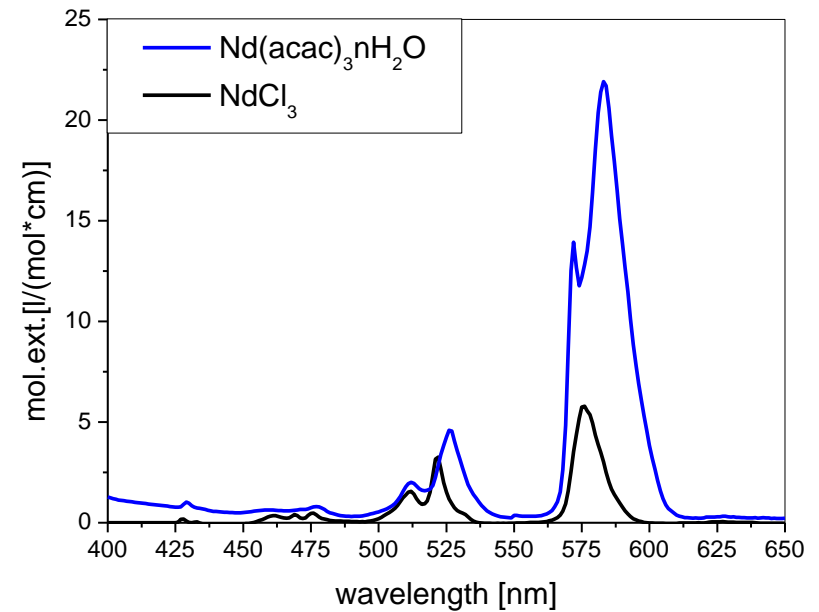
INR/LNGS (ca. 5 g/l):

Tests in
2 l cell

Light yield
measurements



MPIK (2003): Tests on BDK and carboxylate versions (F.X.Hartmann et al.)



R&D in US for SNO+



Summary

- Many solvent and fluor candidates
 - Choice depends on application and detector characteristics
 - Requirement for „safe“ scintillators (PXE, LAB,...)
- Energy transfer models allow light yield predictions
- Several applications for metal loaded scintillators
- Liquid scintillators were / will be key technology for large scale neutrino detectors
- Significant improvement in last years (stability etc.)

